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Solute. what is dissolved.
Solvent. what does the dissolving.
Soluble. can be dissolved. Tyndall
Effect. dispersal of particles; ex.
seeing beams of light through fog
(scattering light) Solutions.

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Capillary Action. Fluid. Surface

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Review Answer: the attraction of the surface of a liquid to the surface of a.... a substance that can flow and therefore take the shape of its.... a force that tends to pull adjacent parts of a liquid's surfac....

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Homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a

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Mixtures The Solution Process
Concentrations of Solution 1 2
Section 12.1 Types of Mixtures
Chapter 12 Section 1 Types of
Mixtures p. 401-406 2 3
Vocabulary Colloid Dispersed
phase Dispersing medium
Electrolyte Nonelectrolyte Solute

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Teacher Notes and Answers

Chapter 12 SECTION 1 SHORT

ANSWER 1. c 2. a 3. b 2. a.

alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with

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those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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space provided, write the letter of
the term or phrase that best
completes each statement or best
answers each question. _____ 1.
Agitation prevents settling in a(n)
a. alloy. b. homogeneous mixture.

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c. suspension. d. gaseous mixture. ____ 2. All of the following are heterogeneous ...

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NCERT INTEXT QUESTIONS. 13.1.
Classify the following amines as
primary, secondary and tertiary:

Ans: (i) 1° (ii) 3° (iii) 1° (iv) 2°

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13.2. Write the structures of different isomeric amines corresponding to the molecular formula, $C_4H_{11}N$. (i) Write the IUPAC names of all the isomers

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amount in moles of the
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