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when the same number of each element is represented on the
reactant and product sides. Equations must be balanced to
accurately reflect the law of conservation of matter. 3. Answer Key
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CHAPTER 7 REVIEW Chemical Formulas and Chemical
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Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H_2SO_3 6 b. S in MgSO_4 2 c. S in K_2S 1 d. Cu in Cu_2S 6 e. Cr in Na_2CrO_4 5 f. N in HNO_3 4 g. C in $(\text{HCO}_3)_3$ 3 h. N in $(\text{NH}_4)_2\text{SO}_4$ 2 a.

~~7 Chemical Formulas and Chemical Compounds~~

Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company ISBN 978-0-07802-152-7

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Answer: (b) Lewis Base. Explanation: R_2NH (Amines) behaves as a Lewis base because it is capable of donating a lone pair of electron. We hope the given NCERT MCQ Questions for Class 11 Chemistry Chapter 7 Equilibrium with Answers Pdf free download will help you.

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As shown in Figure 7.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium(III) oxide forms. We can describe this reaction with a chemical equation An expression that gives the identities and quantities of the ...

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7/18/2019 chap 1-16 140/245 Chapter 10 Conjugation in Alkadienes and Allylic Systems: Answers Prof. Sivaguru Jayaraman Page Chem 341: Organic Chemistry, Prof. Sivaguru Jayaraman, Fall 2010. & 16. What is the kinetically controlled

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product in the following reaction? A) A B) B C) C D) D Ans: A 17.
Rank the following carbocations in decreasing order of stability.

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